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CARBENE COMPLEXES

IV*. FAR INFRARED AND ^{31}P NMR SPECTRA OF PALLADIUM AND PLATINUM CARBENE COMPLEXES

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Summary

Infrared bands mainly associated with $\nu(\text{M}-\text{X}_2)$ stretching modes ($\text{M} = \text{Pd}$ or Pt , and $\text{X} = \text{Cl}$, Br , or P) have been identified in the spectra of 35 carbene complexes. Based on these results and on $|^1J|$ ($^{31}\text{P}-^{195}\text{Pt}$) the *trans*-influence of the carbene ligands is assessed.

Introduction

A series of palladium and platinum carbene complexes of the ligand 1,3-dimethylimidazolidine-2-ylidene and related carbenes has been prepared. Preparative details [2, 3], chemical properties [4, 5], IR, ^1H NMR [5], and ^{13}C NMR [1] spectra have been published. We now report other spectroscopic studies (far IR and ^{31}P NMR) which in addition to providing structural information, afford further data concerning the *trans*-influence of the new carbene ligands.

Infrared spectra

Carbene complexes of the series *cis*- or *trans*- L_2MCl_2 , where $\text{L} =$ tertiary phosphine or arsine and $\text{M} = \text{Pd}$ or Pt , show strong infrared absorptions in the range $270\text{--}350\text{ cm}^{-1}$ (see Table 1). These bands are tentatively attributed to $\nu(\text{M}-\text{Cl}_2)$ vibrational modes. In order to confirm this assignment a number of analogous bromo complexes was prepared: these (*cis*-isomers) showed strong absorptions at $175\text{--}205\text{ cm}^{-1}$ [$\nu(\text{M}-\text{Br}_2)$] and the bands previously assigned to metal-chlorine vibrations were absent (see Tables 1 and 2). However, bands associated with $\nu(\text{M}-\text{Br}_2)$ were not sufficiently strong for most of the *trans*-

* For part I see ref. 3; part II, ref. 5; part III, ref. 1.

TABLE 1

METAL-CHLORINE STRETCHING VIBRATIONS IN THE SERIES *cis*- AND *trans*-L₂MCl₂ (L = TERTIARY PHOSPHINE OR ARSINE, OR CARBENE, M = Pd OR Pt)

Formula	<i>trans</i> -Series		<i>cis</i> -Series	
	Compound	$\nu(\text{M}-\text{Cl}_2)^a$	Compound	$\nu(\text{M}-\text{Cl}_2)^a$
PtCl ₂ (CNPhCH ₂ CH ₂ NPh)(PEt ₃)	(I)	341 s	(II)	308 s, 277 s
PtCl ₂ (CNPhCH ₂ CH ₂ NPh)(PBu ₃)	(III)	337 s	(IV)	305 s, 275 m
PtCl ₂ (CNPhCH ₂ CH ₂ NPh)(PMe ₂ Ph)	(V)	340 s	(VI)	312 s, 285 s
PtCl ₂ (CNPhCH ₂ CH ₂ NPh)(AsEt ₃)	(VII)	340 vs	(VIII)	312 s, 285 s
PtCl ₂ (CNMeCH ₂ CH ₂ NMe)(PEt ₃)	(IX)	339 vs	(X)	312 vs, 288 vs
PtCl ₂ (CNMeCH ₂ CH ₂ NMe)(PPr ₃)	(XI)	340 vs	(XII)	359 vs, 286 s
PtCl ₂ (CNMeCH ₂ CH ₂ NMe)(PBu ₃)	(XIII)	340 vs	(XIV)	308 vs, 285 s
PtCl ₂ (CNMeCH ₂ CH ₂ NMe)(PMe ₂ Ph)	(XV)	b	(XVI)	311 s, (305 sh), 283 s
PtCl ₂ (CNMeCH ₂ CH ₂ NMe)(AsEt ₃)	(XVII)	338 vs	(XVIII)	310 vs, 295 vs
PtCl ₂ (CNMeC ₆ H ₄ S-o)(PEt ₃)	(XIX)	347 s (320 sh)	(XX)	310 s, (302 sh), 281 s
PtCl ₂ (CNMeC ₆ H ₄ S-o)(PBu ₃)	(XXI)	348 s, 325 s ^c	(XXII)	310 s, (302 sh), 285 s
PtCl ₂ (CNMeC ₆ H ₄ S-o)(PMe ₂ Ph)	(XXIII)	b	(XXIV)	315 s, (310 sh), 285 s
PdCl ₂ (CNPhCH ₂ CH ₂ NPh)(PEt ₃)	(XXV)	351 s (318 m)	(XXVI)	302 s, 272 s
PdCl ₂ (CNPhCH ₂ CH ₂ NPh)(PBu ₃)	(XXVII)	350 s	(XXVIII)	306 s, 275 m
PdCl ₂ (CNMeCH ₂ CH ₂ NMe)(PEt ₃)	(XXIX)	b	(XXX)	309 s, (302 sh), 285 s
PdCl ₂ (CNMeCH ₂ CH ₂ NMe)(PBu ₃)	(XXXI)	b	(XXXII)	305 s, 292 s
PdCl ₂ (CNMeC ₆ H ₄ S-o)(PEt ₃)	(XXXIII)	b	(XXXIV)	311 s, (303 sh), 280 s
PdCl ₂ (CNMeC ₆ H ₄ S-o)(PBu ₃)	(XXXV)	b	(XXXVI)	312 s, (307 sh), 447 s

^a Spectra run as Nujol mulls using KBr discs. ^b Data not available; *trans* → *cis* isomerisation too facile. ^c Also shoulders in spectrum at 330 and 320 cm⁻¹. All peaks in the appropriate region are quoted, but the shoulders (in parentheses) are not necessarily associated with $\nu(\text{M}-\text{Cl}_2)$.

isomers that any *reliable* assignments could be made, and, for these, figures have been omitted from Table 1. In common with most other square planar dihalides the *trans*-isomers in the present compounds show only one strong $\nu(\text{M}-\text{Cl}_2)$ band whereas there are invariably two major absorptions associated with $\nu(\text{M}-\text{X}_2)$ in the *cis*-series.

Far infrared frequencies, and especially $\nu(\text{M}-\text{Cl}_2)$ have also been useful in assessment of ligand *trans*-influence [6]. In the present series we have already noted [2, 4], based on X-ray structural data, that the carbenes exert a *trans*-influence for platinum(II) very close to that of tertiary phosphines. Infrared spectra, however, are less simple to interpret for this series, in which the *cis*-complexes are *unsymmetrical*. However, if instead of using mean $\nu(\text{Pt}-\text{X}_2)$ values (see e.g., ref. 6) we consider both frequencies a trend is clearly discernible; e.g., compare [(compound), $\nu(\text{Pt}-\text{Cl}_2)$ cm⁻¹]: (I) 308, 277; (IX) 312, 288; and (XIX) 310, 281; with *cis*-PtCl₂(PEt₃)₂ 305, 283 [7]. The values clearly suggest a very similar *trans*-influence for the three carbenes and triethylphosphine. Analogous

TABLE 2
METAL-BROMINE STRETCHING VIBRATIONS IN THE SERIES *cis*- AND *trans*-L₂MBr₂

Formula	<i>trans</i> -Series		<i>cis</i> -Series	
	Compound	$\nu(\text{M}-\text{Br}_2)^a$	Compound	$\nu(\text{M}-\text{Br}_2)^a$
PtBr ₂ (CNPhCH ₂ CH ₂ NPh)(PEt ₃)	(XXXVII)	248 m	(XXXVIII)	195 vs. 203 s
PtBr ₂ (CNMeCH ₂ CH ₂ NMe)(PEt ₃)	(XXXIX)	b	(XL)	192 vs. 203 s
PtBr ₂ (CNMeC ₆ H ₄ S-o)(PEt ₃)	(XLI)	b	(XLII)	197 s, 187 s
PdBr ₂ (CNPhCH ₂ CH ₂ NPh)(PEt ₃)	(XLIII)	b	(XLIV)	b
PdBr ₂ (CNPhCH ₂ CH ₂ NPh)(PBu ₃)	(XLV)	b	(XLVI)	197, 179 s

^a Samples ground with polythene, and spectra run on polythene-based discs. ^b Bands not sufficiently well resolved to make reliable assignments.

comparisons may be made for other groups of compounds, including the bromo and palladium complexes.

Metal-phosphorus vibrations for the *trans*-compounds are not subject to the same difficulties, and are shown in Table 3, together with metal-arsenic data. However in this case coupling, of $\nu(\text{M}-\text{P})$ or $\nu(\text{M}-\text{As})$ with internal modes of the ligand [e.g., $\delta(\text{PCC})$], is a serious drawback [8].

In the present context assignment of the bands having largely $\nu(\text{M}-\text{P})$ character is relatively straightforward. Thus we have an extensive series of closely related derivatives, for which arsine analogues are available for comparison,

TABLE 3
METAL-PHOSPHORUS AND -ARSENIC VIBRATIONS IN THE SERIES *cis*- AND *trans*-L₂MX₂ (L = TERTIARY PHOSPHINE OR ARSINE, OR CARBENE; X = Cl or Br; M = Pd or Pt)

<i>trans</i> -Series		<i>cis</i> -Series	
Compound	$\nu(\text{M}-\text{P})$ or $\nu(\text{M}-\text{As})^a$	Compound	$\nu(\text{M}-\text{P})$ or $\nu(\text{M}-\text{As})^a$
(I)	422 w	(II)	439 m
(III)	455 w	(IV)	475 w
(V)	442 m	(VI)	452 w
(VII)		(VIII)	306 w
(IX)	421 m	(X)	435 m
(XI)	443 m	(XII)	447 s
(XIII)	445 w	(XIV)	442 s
(XV)		(XVI)	437 s
(XVII)	436 w	(XVIII)	440 m
(XIX)	435 s, 425 sh	(XX)	435 m, 430 sh
(XXI)	435 s	(XXII)	430 s
(XXIII)		(XXIV)	437 w
(XXV)	415 w	(XXVI)	435 w
(XXVII)		(XXVIII)	472 w
(XXIX)		(XXX)	435 w, 425 w
(XXXI)		(XXXII)	440 m
(XXXIII)		(XXXIV)	430 s
(XXXV)		(XXXVI)	447 s
(XXXVII)	420 s	(XXXVIII)	438 w
(XXXIX)	432 m	(XL)	447 s
(XLI)	431 s	(XLII)	437 m
(XLIII)	412 w	(XLIV)	412 m
(XLV)	452 w	(XLVI)	468 w

^a Spectra run as nujol mulls between KBr discs.

and the region [7, 9] of the spectrum for $\nu(\text{M}-\text{P})$ exhibits only a single band. It should be pointed out that the free ligands also exhibit a single weak band (unassigned) in this region; however, this band is not present in metal complexes which do not contain phosphines, and moreover the present complexes show but a single absorption within the $\nu(\text{M}-\text{P})$ range. As a means of assessing *trans*-influence, $\nu(\text{M}-\text{P})$ data are clearly unreliable, but at least do not conflict with our general conclusion; note: (I) has $\nu(\text{Pt}-\text{P})$ at 422 cm^{-1} , *trans*- $\text{PtCl}_2(\text{PEt}_3)_2$, at 419 cm^{-1} [9].

$\nu(\text{M}-\text{C})$ has proved difficult to assign in complexes of the present type. All these compounds exhibit a medium-weak band in the region $500\text{--}550\text{ cm}^{-1}$ which may be associated with the $\text{M}-\text{C}_{\text{carb}}$ stretching mode, but the parent olefins also show a band in the same region. Moreover, the Pt^{IV} complexes $\text{PtCl}_4(\text{CHNMe}_2)_2\text{PEt}_3$ and $\text{PtCl}_4(\text{CHNMe}_2)_2\text{PBu}_3$, show no absorptions between 480 and 600 cm^{-1} [10]. Those in the present series at $510\text{--}540\text{ cm}^{-1}$ may therefore arise from internal modes of the carbene ligands.

^{31}P NMR spectra

Data for the carbene complexes are presented in Table 4. The *cis*-complexes invariably exhibit larger $|^1J|(^{31}\text{P}-^{195}\text{Pt})$, which confirms other stereochemical assignments. Variations in $J(\text{P}-\text{Pt})$ in the *trans*-complexes can be interpreted in terms of the appropriate form of the Pople and Santry expression for coupling [11], and can be used as a measure of the *trans*-influence of the ligand *trans* to P. In this procedure a high $J(^{31}\text{P}-^{195}\text{Pt})$ implies a low *trans*-influence for the appropriate ligand. Once again we note the general similarity to tertiary phosphines, [compare $|^1J|(^{31}\text{P}-^{195}\text{Pt})$ for (I), 2.44 kHz , with that of *trans*- $\text{PtCl}_2(\text{PBu}_3)_2$, 2.39 kHz], but the method is sensitive enough to make distinctions possible between the different carbene ligands. We should thus have the *trans*-influence order: $\text{Et}_3\text{P} \approx 1,3\text{-dimethylimidazolidinylidene} > N\text{-methylbenzothiazolinyldiene} \geq 1,3\text{-dimethylimidazolidinylidene}$:

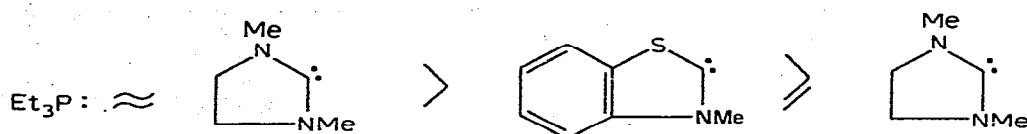


TABLE 4

^{31}P CHEMICAL SHIFTS AND $|J|(^{31}\text{P}-^{195}\text{Pt})$ FOR THE SERIES *cis*- AND *trans*- L_2MX_2 (L = TERTIARY PHOSPHINE OR CARBENE; X = Cl or Br; M = Pd or Pt).

<i>trans</i> -Series			<i>cis</i> -Series		
Compound	$\delta^{31}\text{P}^b$	$ J (^{31}\text{P}-^{195}\text{Pt})^a$	Compound	$\delta^{31}\text{P}^b$	$ J (^{31}\text{P}-^{195}\text{Pt})^c$
(I)	+110.8	2.44	(II)	<i>d</i>	
(IX)	+110.0	2.35	(X)	+136.2	3.72
(XI)	+119.1	2.34	(XII)	+114.8	3.75
(XIII)	+ 80.2	2.35	(XIV)	+ 46.3	3.26
(XIX)	+107.2	2.44	(XX)	<i>d</i>	
(XXXVII)	+114	2.47	(XXXVIII)	<i>d</i>	
(XXXIX)	+121	2.36	(XL)	+114	4.08

^a Measured in chloroform solution. ^b In ppm from external (capillary) P_4O_6 . ^c In kHz. ^d *cis*-Isomer too insoluble for measurements.

Experimental

The preparations of the complexes have been previously reported [2, 3]. IR spectra were recorded using a Perkin Elmer model 457 grating instrument, and ^{31}P NMR spectra were obtained using a Perkin Elmer model R10 spectrometer operating a 24.29 MHz, with a Northern Electronics C.A.T. Far infrared spectra were measured using an R.I.I.C. FS 620 Fourier transform spectrometer, and samples were made up as polythene discs.

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